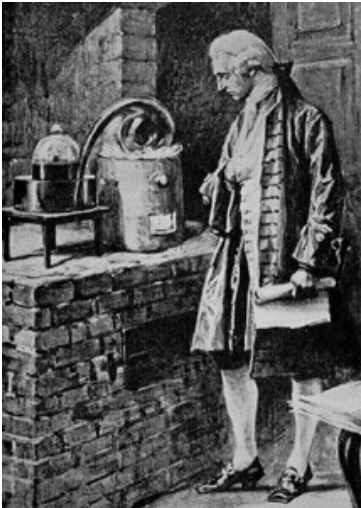


**פרופסור צבי מזא"ה**

## חוק שימור החומר: מסה אינה נעלמת או נוצרת

Lavoisier (1743-1794):



Joseph-Louis Lagrange, remarked of this event, "It took them only an instant to cut off that head, and a hundred years may not produce another like it."

**James Prescott Joule (1818 – 1889) :**

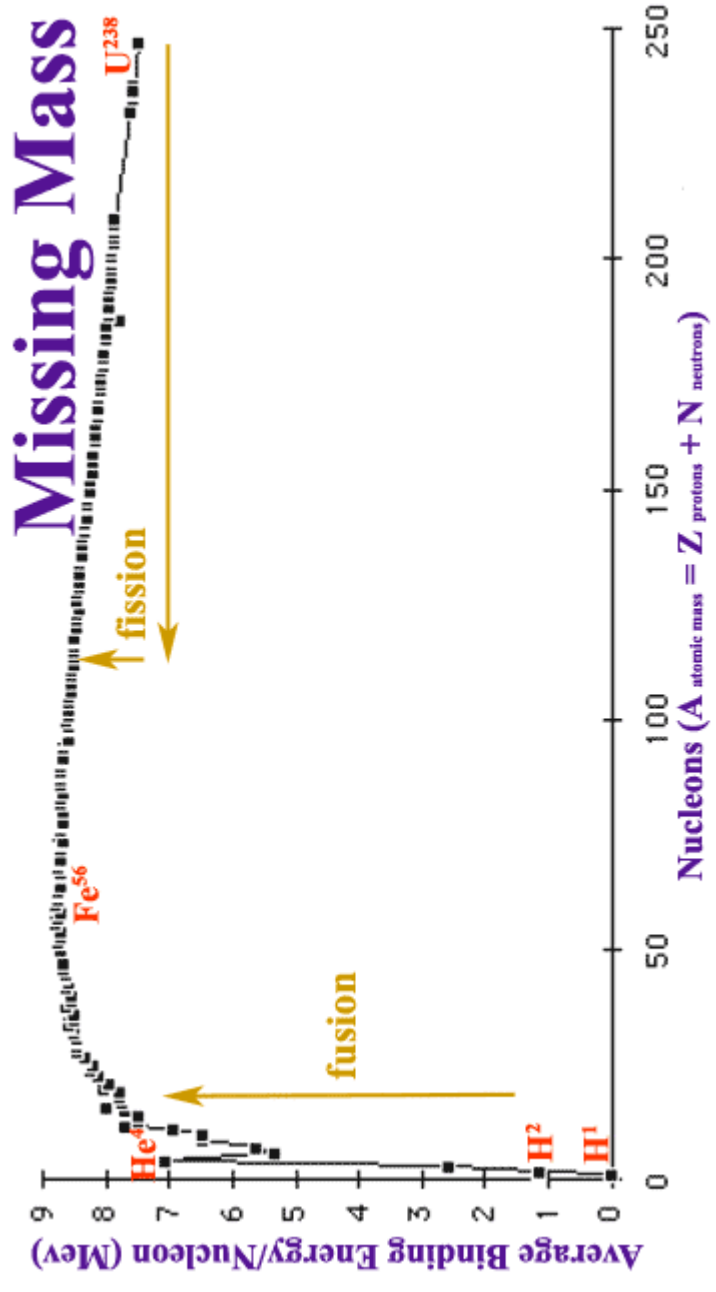


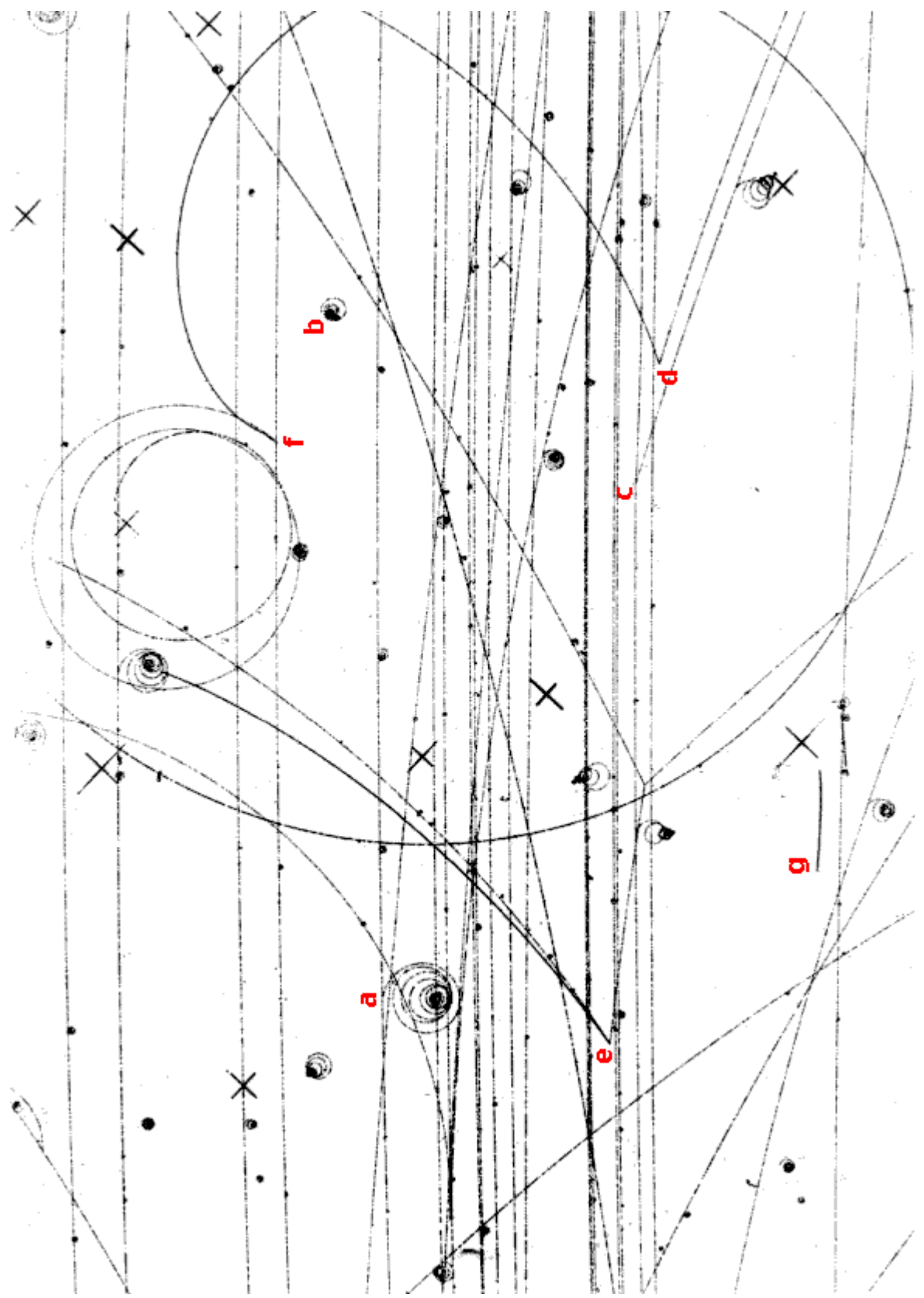
חוק שימור האנרגיה

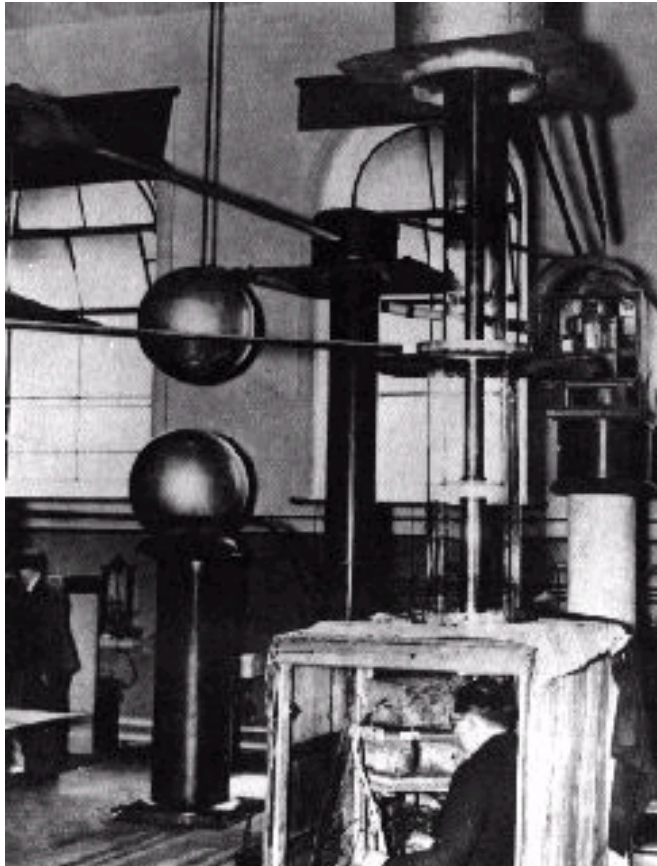
# Periodic Table of the Elements

IA																					0																																																					
1	IIA												IIIA	IVA	VA	VIA	VIIA	0																																																								
1	<b>H</b>	<b>He</b>											<b>Li</b>	<b>Be</b>											<b>B</b>	<b>C</b>	<b>N</b>	<b>O</b>	<b>F</b>	<b>Ne</b>																																												
2	<b>Li</b>	<b>Be</b>											<b>B</b>	<b>C</b>	<b>N</b>	<b>O</b>	<b>F</b>	<b>Ne</b>																																																								
3	<b>Na</b>	<b>Mg</b>											<b>Al</b>	<b>Si</b>	<b>P</b>	<b>S</b>	<b>Cl</b>	<b>Ar</b>																																																								
4	<b>K</b>	<b>Ca</b>	IIIB	IVB	VB	VIB		VII		VIIB	IB	IB	IIIB	IVB	VB	VIIB	VIIB	IIIB	IVB	VB	VIIB	VIIB	IIIB	IVB	VB	VIIB	VIIB	IIIB	IVB	VB	VIIB	VIIB	IIIB	IVB	VB	VIIB	VIIB																																					
5	<b>Rb</b>	<b>Sr</b>	<b>Y</b>	<b>Zr</b>	<b>Nb</b>	<b>Mo</b>	<b>Tc</b>	<b>Ru</b>	<b>Rh</b>	<b>Pd</b>	<b>Ag</b>	<b>Cd</b>	<b>In</b>	<b>Sn</b>	<b>Sb</b>	<b>Te</b>	<b>I</b>	<b>Xe</b>																																																								
6	<b>Cs</b>	<b>Ba</b>	* <b>La</b>	<b>Hf</b>	<b>Ta</b>	<b>W</b>	<b>Re</b>	<b>Os</b>	<b>Ir</b>	<b>Pt</b>	<b>Au</b>	<b>Hg</b>	<b>Tl</b>	<b>Pb</b>	<b>Bi</b>	<b>Po</b>	<b>At</b>	<b>Rn</b>																																																								
7	<b>Fr</b>	<b>Ra</b>	+ <b>Ac</b>	<b>Rf</b>	<b>Ha</b>	<b>Sg</b>	<b>Ns</b>	<b>Hs</b>	<b>Mt</b>	<b>110</b>	<b>111</b>	<b>112</b>	<b>113</b>	<b>114</b>	<b>115</b>	<b>116</b>	<b>117</b>	<b>118</b>																																																								
			<table border="1"> <tr> <td>58</td> <td><b>Ce</b></td> <td>59</td> <td><b>Pr</b></td> <td>60</td> <td><b>Nd</b></td> <td>61</td> <td><b>Pm</b></td> <td>62</td> <td><b>Sm</b></td> <td>63</td> <td><b>Eu</b></td> <td>64</td> <td><b>Gd</b></td> <td>65</td> <td><b>Tb</b></td> <td>66</td> <td><b>Dy</b></td> <td>67</td> <td><b>Ho</b></td> <td>68</td> <td><b>Er</b></td> <td>69</td> <td><b>Tm</b></td> <td>70</td> <td><b>Yb</b></td> <td>71</td> <td><b>Lu</b></td> </tr> <tr> <td>90</td> <td><b>Th</b></td> <td>91</td> <td><b>Pa</b></td> <td>92</td> <td><b>U</b></td> <td>93</td> <td><b>Np</b></td> <td>94</td> <td><b>Pu</b></td> <td>95</td> <td><b>Am</b></td> <td>96</td> <td><b>Cm</b></td> <td>97</td> <td><b>Bk</b></td> <td>98</td> <td><b>Cf</b></td> <td>99</td> <td><b>Es</b></td> <td>100</td> <td><b>Fm</b></td> <td>101</td> <td><b>Md</b></td> <td>102</td> <td><b>No</b></td> <td>103</td> <td><b>Lr</b></td> </tr> </table>																58	<b>Ce</b>	59	<b>Pr</b>	60	<b>Nd</b>	61	<b>Pm</b>	62	<b>Sm</b>	63	<b>Eu</b>	64	<b>Gd</b>	65	<b>Tb</b>	66	<b>Dy</b>	67	<b>Ho</b>	68	<b>Er</b>	69	<b>Tm</b>	70	<b>Yb</b>	71	<b>Lu</b>	90	<b>Th</b>	91	<b>Pa</b>	92	<b>U</b>	93	<b>Np</b>	94	<b>Pu</b>	95	<b>Am</b>	96	<b>Cm</b>	97	<b>Bk</b>	98	<b>Cf</b>	99	<b>Es</b>	100	<b>Fm</b>	101	<b>Md</b>	102	<b>No</b>	103	<b>Lr</b>
58	<b>Ce</b>	59	<b>Pr</b>	60	<b>Nd</b>	61	<b>Pm</b>	62	<b>Sm</b>	63	<b>Eu</b>	64	<b>Gd</b>	65	<b>Tb</b>	66	<b>Dy</b>	67	<b>Ho</b>	68	<b>Er</b>	69	<b>Tm</b>	70	<b>Yb</b>	71	<b>Lu</b>																																															
90	<b>Th</b>	91	<b>Pa</b>	92	<b>U</b>	93	<b>Np</b>	94	<b>Pu</b>	95	<b>Am</b>	96	<b>Cm</b>	97	<b>Bk</b>	98	<b>Cf</b>	99	<b>Es</b>	100	<b>Fm</b>	101	<b>Md</b>	102	<b>No</b>	103	<b>Lr</b>																																															

\* Lanthanide Series  
+ Actinide Series







Cockroft

1897-1967



Walton

1903-1995

Nobel Prize, 1951:

*"for their pioneer work on the transmutation of atomic nuclei by artificially accelerated atomic particles"*



**Emilio Gino Segrè**

(1905-1989)



**Owen Chamberlain**

(1920-)

Nobel Prize of 1959:

for their discovery of the antiproton





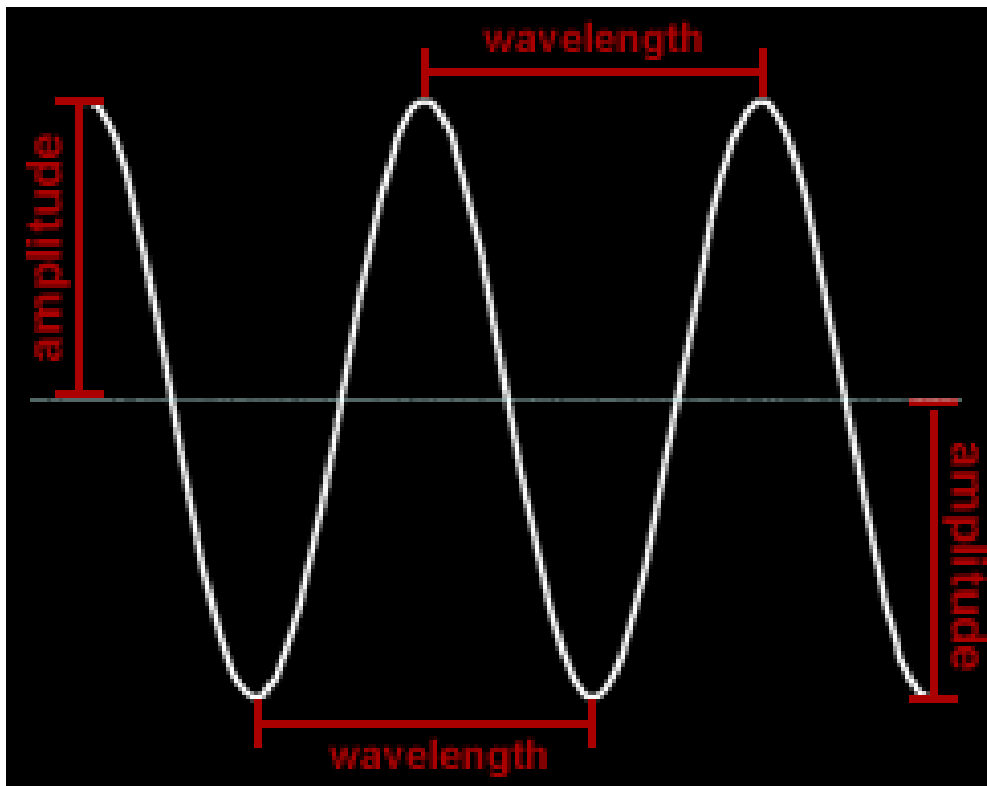
## יצירת האנטי פרוטון

Bevatron

Chamberlain, Segrè,



# Light Waves

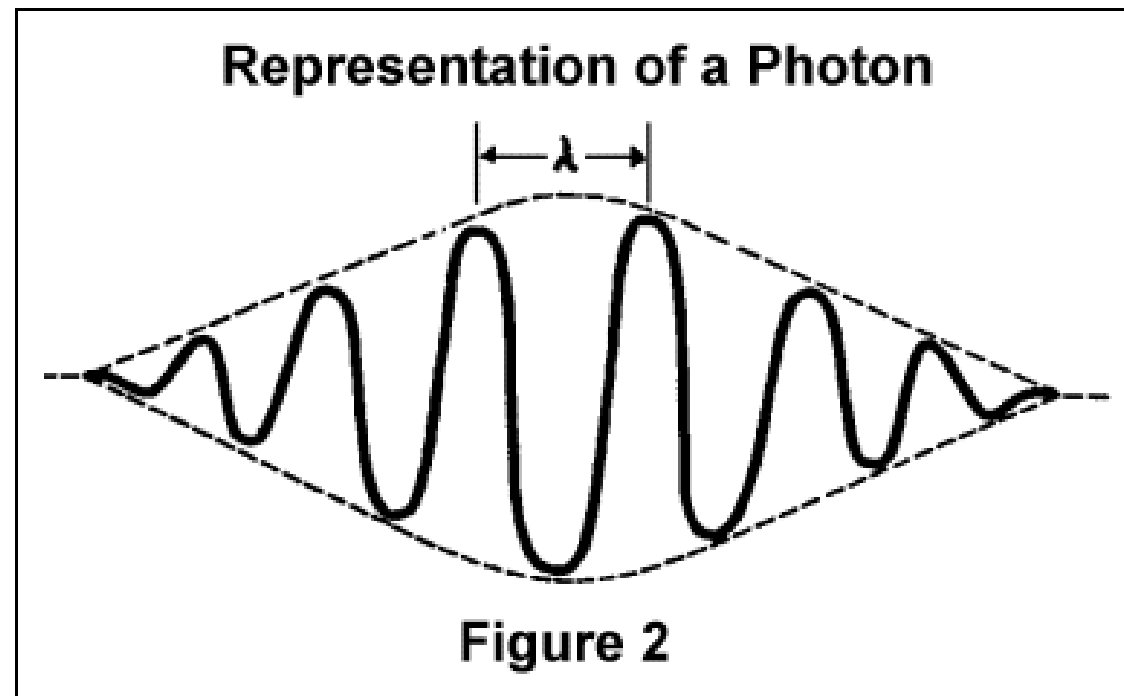


Light waves:

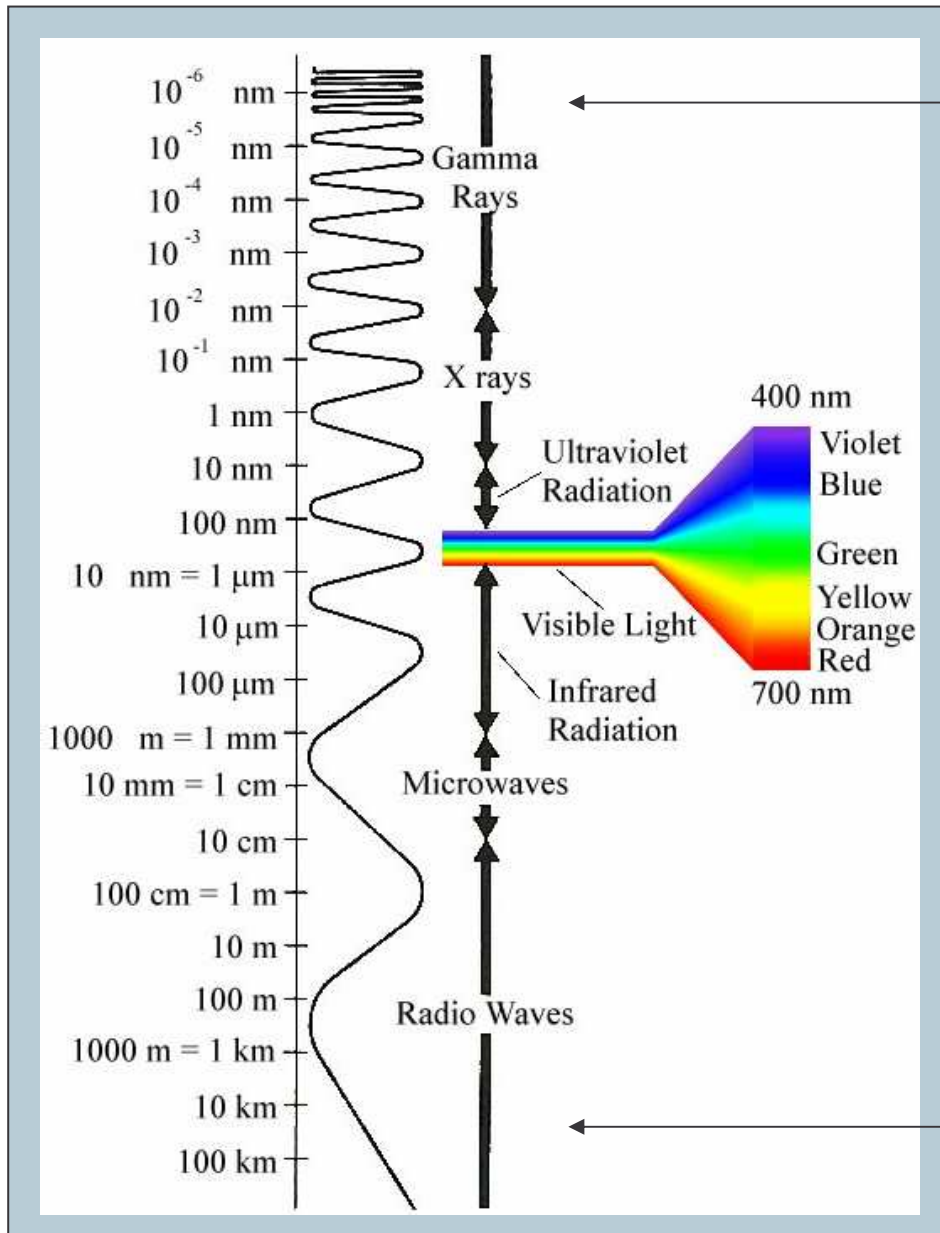
Characterized by:

- Amplitude (A)
- Frequency ( $\nu$ )
- Wavelength ( $\lambda$ )

**Energy  $\propto A^2$**



# הספקטרום האלקטרומגנטי



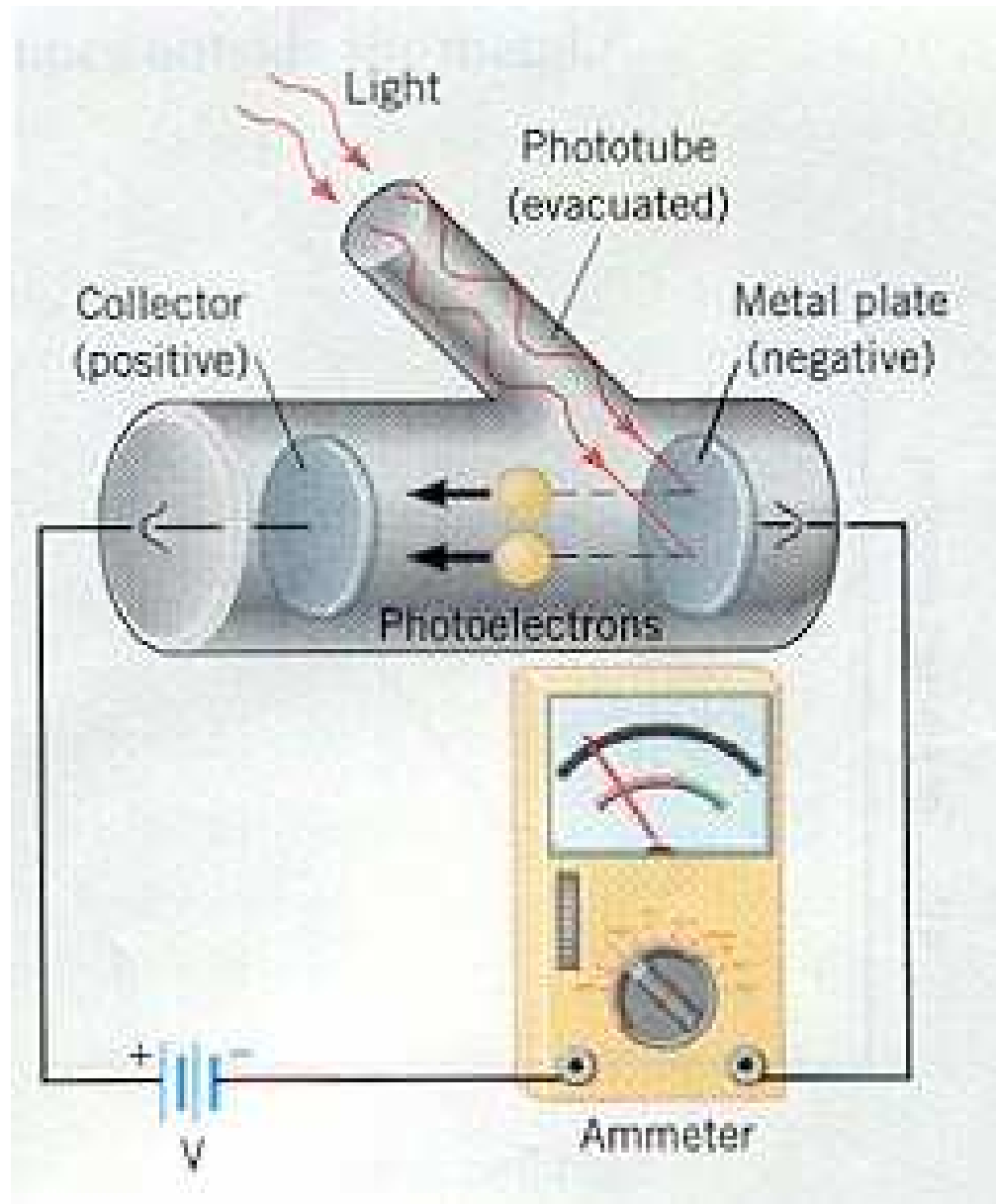
Shortest wavelengths  
(Most energetic photons)

$$E = h \nu = hc/\lambda$$

$h = 6.6 \times 10^{-34} \text{ [J*sec]}$   
(Planck's constant)

Longest wavelengths  
(Least energetic photons)

# האפקט הפוטואלקטרי



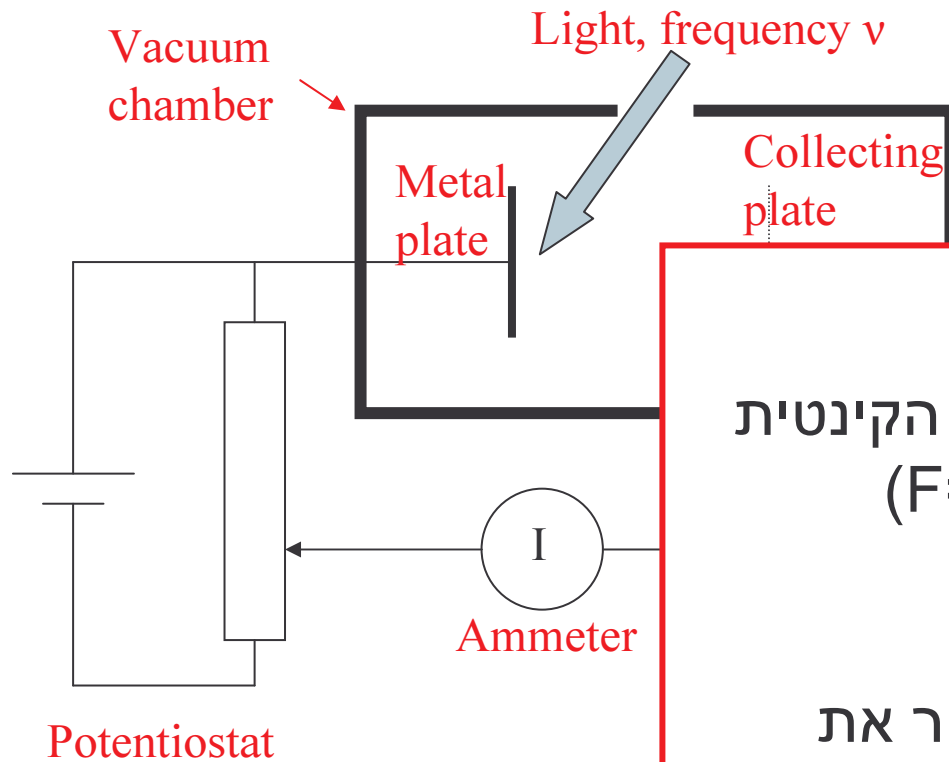
# PHOTOELECTRIC EFFECT

When UV light is shone on a metal plate in a vacuum, it emits charged particles (Hertz 1887), which were later shown to be electrons by J.J. Thomson (1899).

Hertz



J.J. Thomson



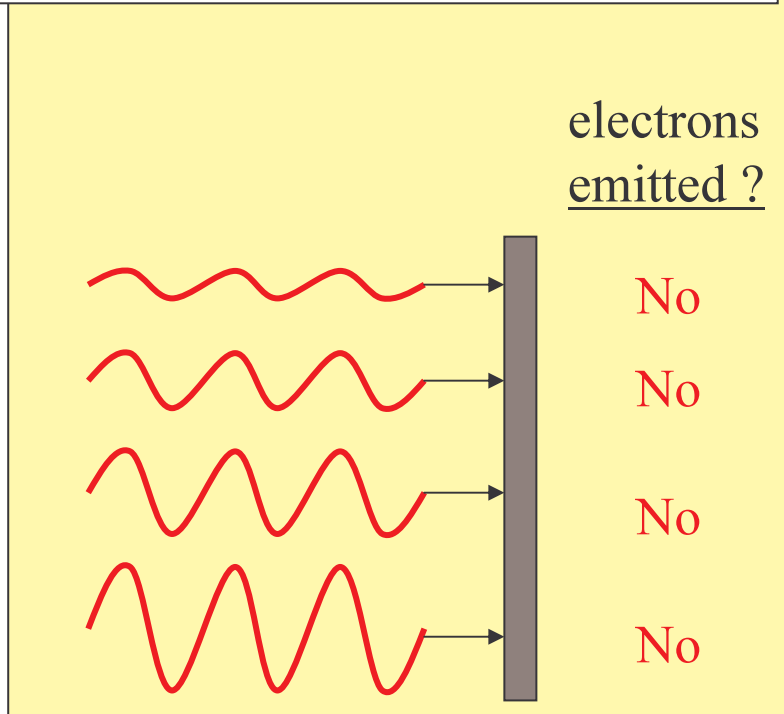
תיאוריה קלסית:

1. עצמת האור תגדיל את האנרגיה הקינטית של האלקטרונים הנפלטים ( $F=eE$ )
2. האפקט קיים לכל תדירות
3. בעצמה נמוכה דרוש זמן כדי ליצור את האפקט

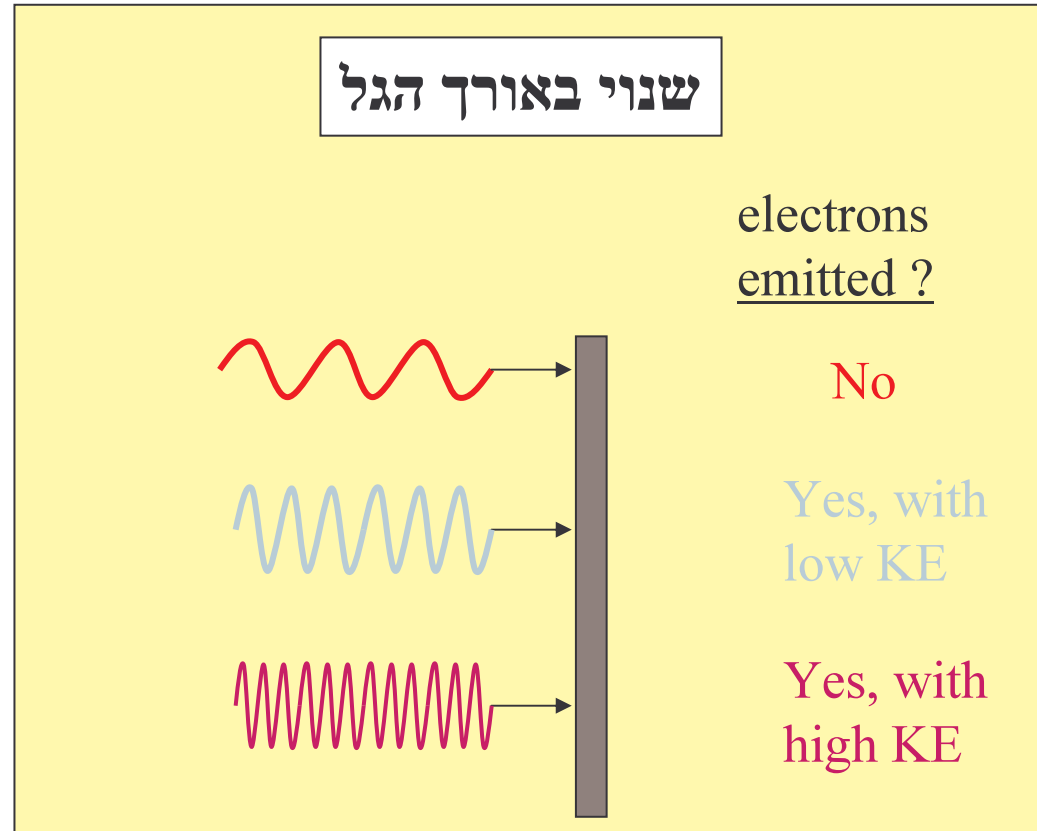
# האפקט הפוטואלקטרי

## השיטה הקלאסית

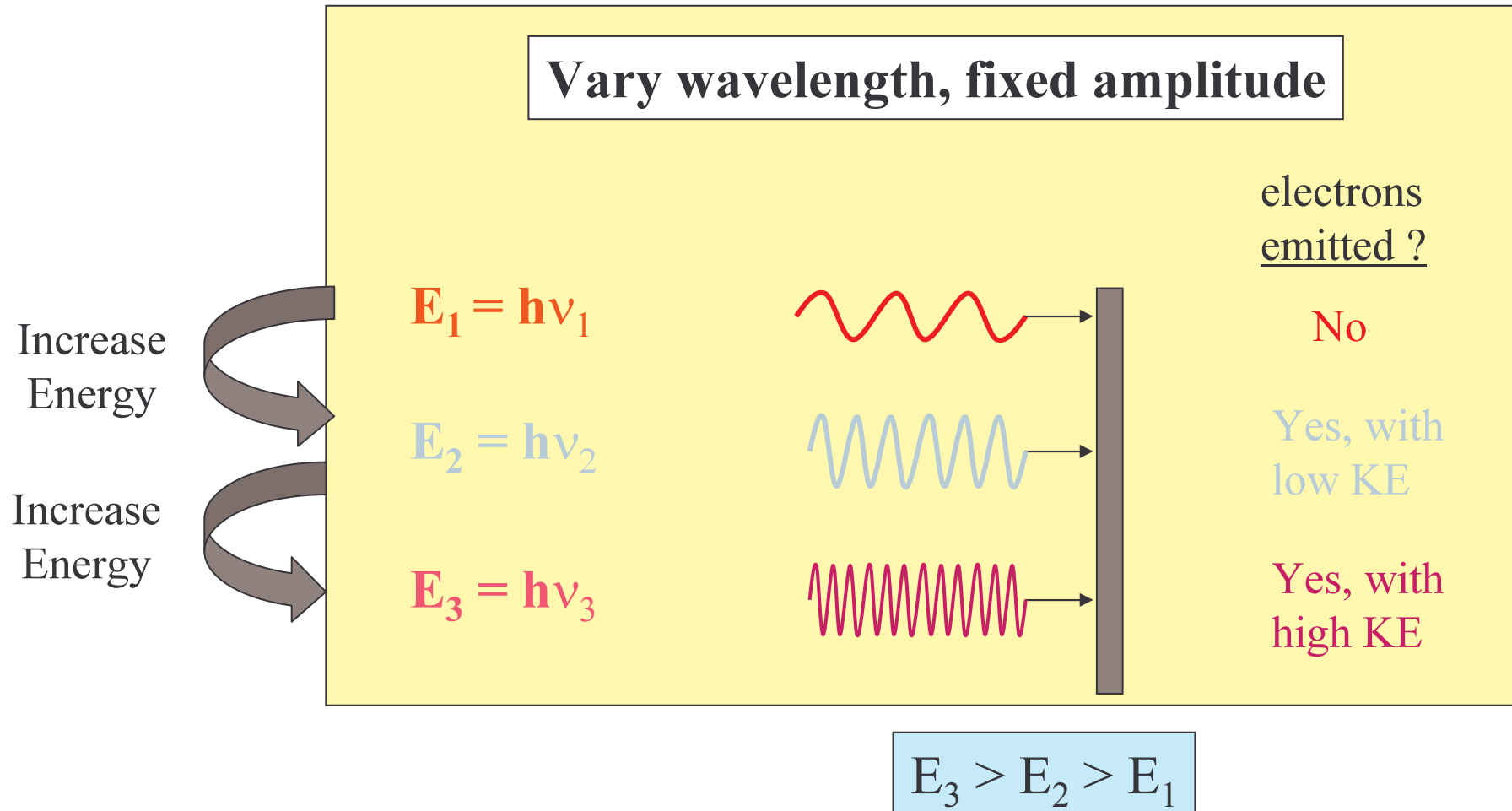
הגדלת האנרגיה על ידי עצמת הגל



שנוי באורך הגל



# ההסבר של הניסוי:





# PHOTOELECTRIC EFFECT (cont)

## Actual results:

Maximum KE of ejected electrons is independent of intensity, but dependent on  $\nu$

For  $\nu < \nu_0$  (i.e. for frequencies below a cut-off frequency) no electrons are emitted

There is no time lag. However, rate of ejection of electrons depends on light intensity.

## Einstein's interpretation (1905):

Light comes in packets of energy (*photons*)

$$E = h\nu$$

An electron absorbs a single photon to leave the material

*Einstein*



*Millikan*



The maximum KE of an emitted electron is then

$$K_{\max} = h\nu - W$$

*Planck constant:*  
universal constant of nature

$$h = 6.63 \times 10^{-34} \text{ Js}$$

*Work function:* minimum energy needed for electron to escape from metal (depends on material, but usually 2-5eV)

Verified in detail through subsequent experiments by Millikan



רוברט מיליקין (1868-1953)

**Nobel Prize, 1923,**

for his work on the elementary charge of  
electricity and on the photoelectric effect.

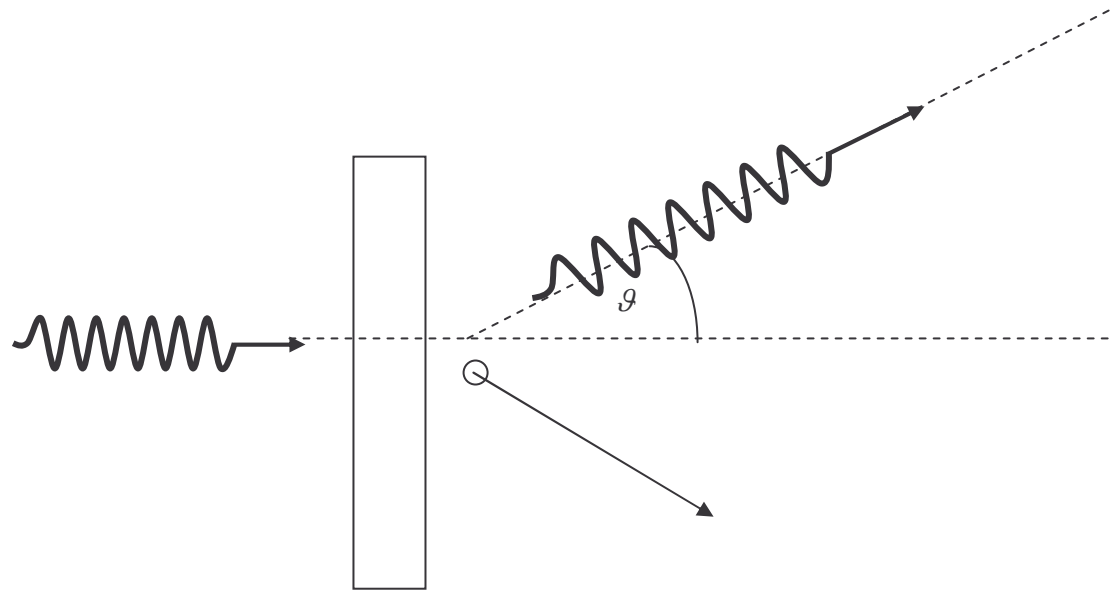


Compton (1892-1962)

Nobel prize, 1927,

for his discovery of the effect  
named after him (performed in 1922)

# אפקט קומפטון

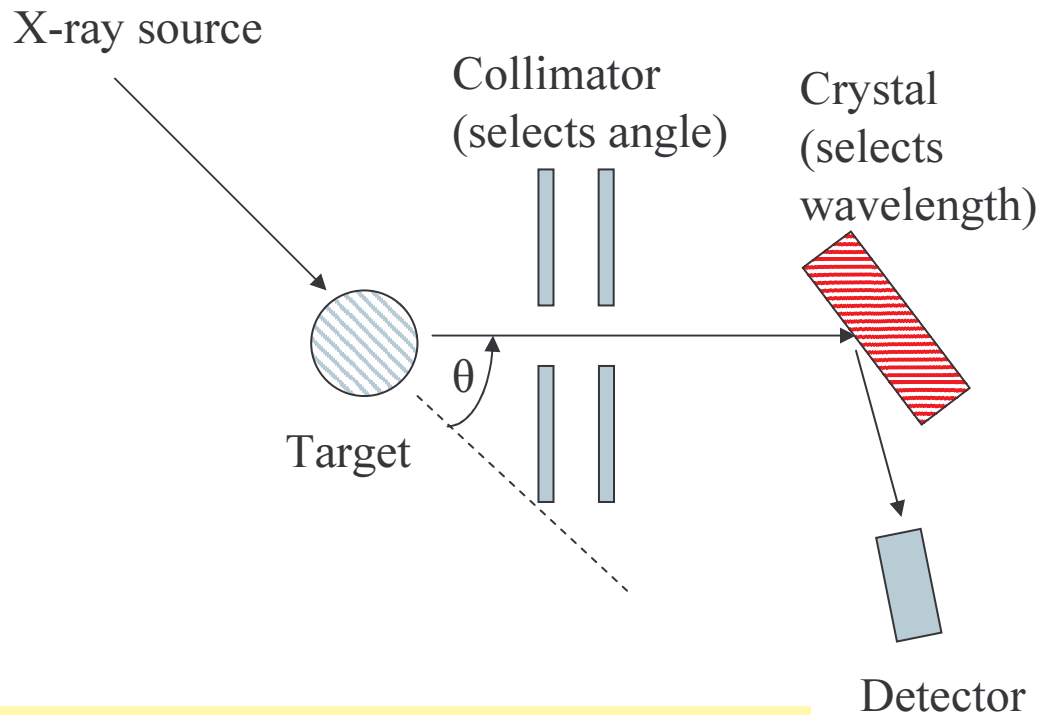


# COMPTON SCATTERING

*Compton*

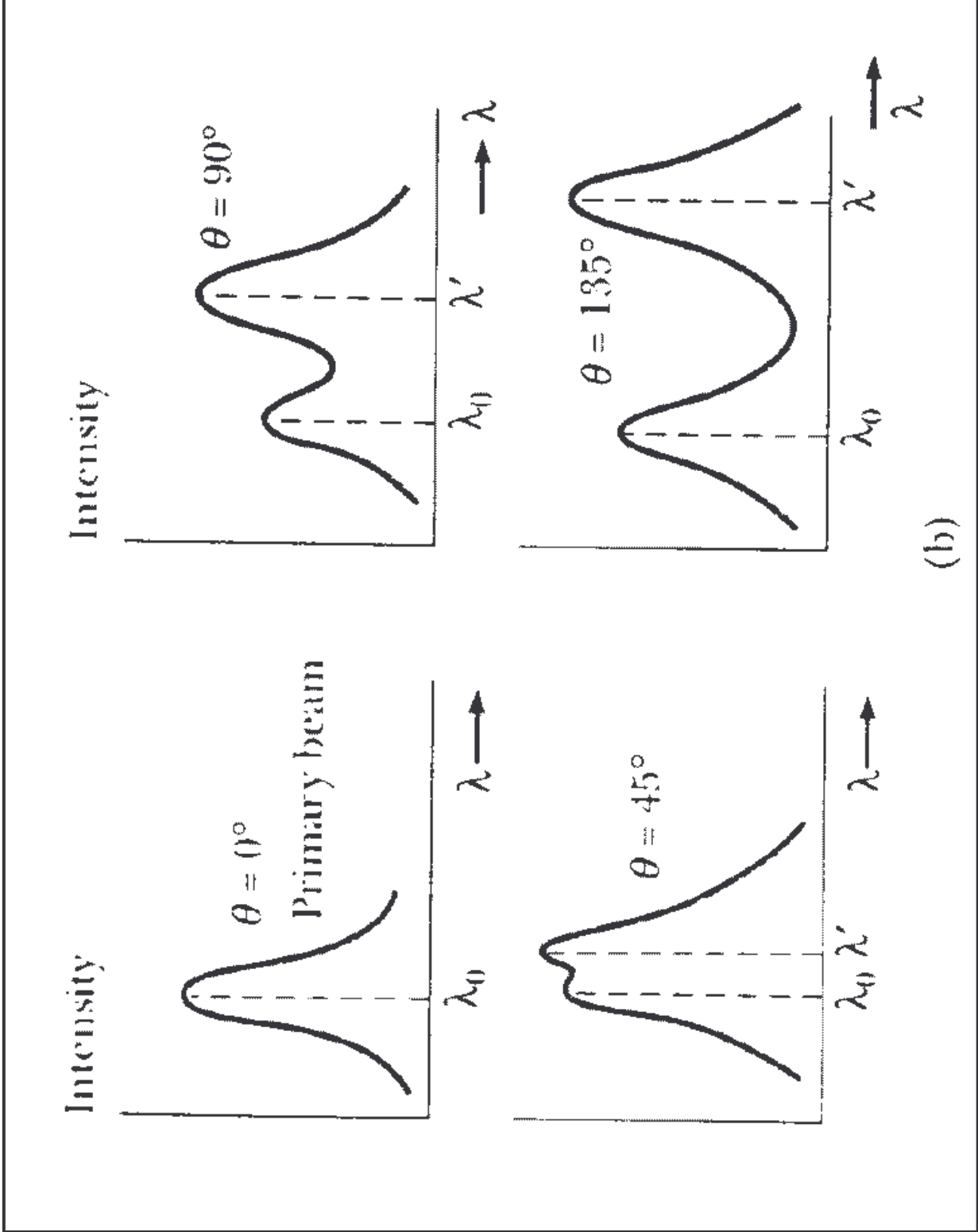


Compton (1923) measured intensity of scattered X-rays from solid target, as function of wavelength for different angles. He won the 1927 Nobel prize.



**Result:** peak in scattered radiation shifts to longer wavelength than source. Amount depends on  $\theta$  (but not on the target material).

A.H. Compton, *Phys. Rev.* **22** 409 (1923)



(b)

# אפקט קומפטון

$$\lambda_{\text{compton}} = \frac{h}{mc}$$

$$\lambda_{\text{compton}}(e) = 0.024 \text{ \AA}$$